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10/790,617	03/01/2004	RameshBabu Boga	KCX-827 (20129)	8844
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

# Application No. Applicant(s) 10/790,617 BOGA ET AL. Office Action Summary Examiner Art Unit JACQUELINE DIRAMIO 1641 -- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --Period for Reply A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS. WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION. Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication. If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication - Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b). Status 1) Responsive to communication(s) filed on 07 August 2008. 2a) ☐ This action is FINAL. 2b) This action is non-final. 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under Ex parte Quayle, 1935 C.D. 11, 453 O.G. 213. Disposition of Claims 4) Claim(s) 1.5-7.12-16 and 18-21 is/are pending in the application. 4a) Of the above claim(s) is/are withdrawn from consideration. 5) Claim(s) \_\_\_\_\_ is/are allowed. 6) Claim(s) 1,5-7,12-16 and 18-21 is/are rejected. 7) Claim(s) \_\_\_\_\_ is/are objected to. 8) Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement. Application Papers 9) The specification is objected to by the Examiner. 10) ☐ The drawing(s) filed on 18 June 2004 is/are: a) ☐ accepted or b) ☐ objected to by the Examiner. Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a). Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d). 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152. Priority under 35 U.S.C. § 119 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some \* c) None of: Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). \* See the attached detailed Office action for a list of the certified copies not received. Attachment(s) 1) Notice of References Cited (PTO-892) 4) Interview Summary (PTO-413)

Notice of Draftsperson's Patent Drawing Review (PTO-948)

Information Disclosure Statement(s) (PTO/SB/08)
 Paper No(s)/Mail Date \_\_\_\_\_\_\_.

Paper No(s)/Mail Date.

6) Other:

5) Notice of Informal Patent Application

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#### DETAILED ACTION

## Status of the Claims

Currently, claims 1, 5-7, 12-16, and 18-21 are pending and under examination

## Withdrawn Rejections

 All previous rejections of the claims under 35 U.S.C. 103(a) are withdrawn in view of Applicant's arguments filed August 7, 2008.

## Response to Arguments

- 3. Applicant's arguments, see p5-6, filed August 7, 2008, with respect to the rejection(s) of the claim(s) under 35 U.S.C. 103(a) have been fully considered and are persuasive. Applicant argues that the MacDonald et al. reference (US 2005/0085739), which was used as a secondary reference in the previous 103(a) rejection, does not qualify as prior art under 130(a) because both the MacDonald et al. reference and the instant application were commonly owned by Kimberly-Clark Worldwide, Inc. at the time the invention was made. This 103(c) statement is found persuasive. Therefore, the rejections have been withdrawn. However, upon further consideration, a new ground(s) of rejection is made and presented below.
- It is noted that no amendments or arguments were made with respect to the 35 U.S.C. 112, second paragraph, rejection previously raised. Therefore, this rejection is maintained below.

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## Claim Rejections - 35 USC § 112

The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

 Claims 12 – 15 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

Claims 12, 14 and 15 recite the term "said fluidic medium," which lacks antecedent basis.

Claims 5-7, 12-16 and 18-21, "an assay device" should be changed to --the assay device-- for clarity.

## NEW GROUNDS OF REJECTION

## Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- Claims 1, 5 and 6 are rejected under 35 U.S.C. 102(b) as being anticipated by Caillouette (US 5,998,161) in light of Sigma-Aldrich Product Catalog.

Caillouette teaches an apparatus (assay device) for detecting the presence or absence of amines in a test sample, said apparatus comprising a porous body (membrane) that defines a color-changing area (detection zone), wherein a

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chemichromic dye is contained within said color-changing area, said chemichromic dye including Bromocresol Green, Bromocresol Purple, or Bromophenol Blue, which are capable of undergoing a detectable color change upon reaction with one or more amines, said chemichromic dye having the general structure recited in Applicant's claim 1 (see Figure 1; and column 1, lines 5-57; and column 2, lines 30-62). It is noted that Bromocresol Green, Bromocresol Purple, and Bromophenol Blue are triarylmethanes as evidenced by Sigma Aldrich (see "Bromocresol Green, Bromocresol Purple, and Bromophenol Blue" located under "Stains and Dyes" within the Product Catalog of Sigma-Aldrich).

With respect to Applicant's claims 5 and 6, the aryl groups of Bromocresol Green, Bromocresol Purple, and/or Bromophenol Blue, which comprise the indicator dyes used by Caillouette, comprise phenyl groups, which are either sulfonic-substituted, alkyl-substituted, or carbonyl-substituted (see "Bromocresol Green, Bromocresol Purple, and Bromophenol Blue" located under "Stains and Dyes" within the Product Catalog of Sigma-Aldrich).

# Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

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The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

- 1. Determining the scope and contents of the prior art.
- 2. Ascertaining the differences between the prior art and the claims at issue.
- 3. Resolving the level of ordinary skill in the pertinent art.
- Considering objective evidence present in the application indicating obviousness or nonobviousness.
- Claims 1, 5, 6, 12 14, 16, 18 and 19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Daniels et al. (US 2006/0008921) in view of Miller et al. (US 7,014,816) and Caillouette (US 5,998,161) in light of Sigma-Aldrich Product Catalog.

Daniels et al. teach an immunochromatographic test strip (assay device) for detecting the presence or absence of an analyte within a test sample, said test strip comprising a porous membrane that is in fluid communication with a detection reagent (probes) conjugated with a specific binding member for the analyte, said porous membrane defining:

a capture region (second detection zone) within which a capture reagent is immobilized to bind to said detection reagent or complexes thereof to generate a detectable (detection) signal, wherein the amount of analyte in the test sample is proportional to the intensity of the detectable signal (see Figures 1 and 3; and paragraphs [0108]-[0120], [0133], [0201]-[0203], and [0232]-[0236]).

Daniels et al. teach the use of their test strip to detect various analytes, including bacteria, viruses and other microorganisms, such as those found in biological fluids, water or food stuffs (see paragraphs [0094], [0095], and [0232]-[0236]). However,

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Daniels et al. fail to teach the detection of amines, wherein the porous membrane includes a first detection zone that comprises an immobilized chemichromic dye in the form of a triarylmethane dye, said dye capable of undergoing a detectable color change upon reaction with one or more amines.

Miller et al. teach a device for detecting amines in a test sample, wherein the device comprises a substrate and a polymeric matrix that contains an indicator compound. The indicator compound can comprise various dyes that are capable of undergoing a detectable color change upon reaction with one or more amines. Amines represent volatile bases that are generated by food decomposition, therefore, the device provides an effective means to indicate the presence of an unwanted biological agent, such as bacteria or fungi, in a sample, particularly a food sample, by colorimetric detection of amines (see Figures; and column 1, lines 20-61; column 3, lines 39-50; column 4, lines 4-49; column 5, lines 8-34; and column 6, lines 5-25).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to include with the device for detecting analytes of Daniels et al. a detection zone that comprises a colorimetric reagent that undergoes a color change in the presence of amines as taught by Miller et al. because Miller et al. teach that a device that comprises an indicator dye that undergoes a color change in the presence of amines provides an effective means to indicate the presence of an unwanted biological agent, such as bacteria or fungi, in a sample, particularly a food sample, by colorimetric detection of amines.

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However, Miller et al. fail to teach that the indicator dye specifically comprises triarylmethane.

Caillouette teaches an apparatus for detecting the presence or absence of amines in a test sample, said apparatus comprising a porous body that defines a color-changing area, wherein an indicator dye is contained within said color-changing area, said indicator dye including Bromocresol Green, Bromocresol Purple, or Bromophenol Blue, which are capable of undergoing a readily detectable color change upon reaction with one or more amines, said indicator dye having the general structure recited in Applicant's claims 1 and 16 (see Figure 1; and column 1, lines 5-57; and column 2, lines 30-62). It is noted that Bromocresol Green, Bromocresol Purple, and Bromophenol Blue are triarylmethanes as evidenced by Sigma Aldrich (see "Bromocresol Green, Bromocresol Purple, and Bromophenol Blue" located under "Stains and Dyes" within the Product Catalog of Sigma-Aldrich).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to substitute the indicator dye in the device of Daniels et al. as modified by Miller et al. with a triarylmethane as taught by Caillouette because Cailloutte teach the benefit of indicator dyes that comprise various chemical compositions, including Bromocresol Green, Bromocresol Purple, and Bromophenol Blue (i.e. triarylmethanes), because these types of indicator dyes are capable of undergoing a readily detectable color change under reaction within one or more amines.

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With respect to Applicant's claims 5, 6, 18, and 20, the aryl groups of Bromocresol Green, Bromocresol Purple, and/or Bromophenol Blue, which comprise the indicator dyes used by Caillouette, comprise phenyl groups, which are either sulfonic-substituted, alkyl-substituted, or carbonyl-substituted (see "Bromocresol Green, Bromocresol Purple, and Bromophenol Blue" located under "Stains and Dyes" within the Product Catalog of Sigma-Aldrich).

 Claims 7 and 20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Daniels et al. (US 2006/0008921) in view of Miller et al. (US 7,014,816) and Caillouette (US 5,998,161), as applied above, and further in view of Horan (US 6,149,952).

The Daniels et al., Miller et al. and Caillouette references discussed in the 103(a) rejection above fail to teach that the indicator dye, or triarylmethane, comprises alphanaphtholbenzein.

Horan teaches a method for determining the presence or absence of contaminating bacteria in a food sample, wherein the method comprises storing a food sample in a package that contains an indicator that is capable of undergoing a calorimetric reaction in the presence of carbonic acid, sulfuric acid, or ammonium hydroxide in order to visibly detect these compounds and thereby indicate the presence or absence (in the case where no reaction occurs) of contaminating bacteria.

Exemplary indicators are taught, wherein indicators such as bromocresol green, bromocresol purple, and p-naphtholbenzein (i.e. alpha-naphtholbenzein) are listed as

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indicators that provide calorimetric responses to the addition of quantities of acid (see Abstract; column 1, lines 49-60; column 2, lines 65-67; column 3, lines 1-12; column 7, lines 48-61; claims 1 and 9).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to substitute the indicator dyes used in the device of Daniels et al., Miller et al. and Caillouette with an alpha-naphtholbenzein as taught by Horan because Horan teaches that p-naphtholbenzein (i.e. alpha-naphtholbenzein) is an equivalent indicator dye to bromocresol green and/or bromocresol purple because of its ability of undergoing a readily detectable calorimetric response in the same environments as bromocresol green and/or bromocresol purple.

 Claims 15 and 21 are rejected under 35 U.S.C. 103(a) as being unpatentable over Daniels et al. (US 2006/0008921) in view of Miller et al. (US 7,014,816) and Caillouette (US 5,998,161), as applied above, and further in view Lawrence et al. (US 6,099,801).

Daniels et al. teach the inclusion of a control region downstream from said capture region, however, Daniels et al., as well as Miller et al. and Caillouette, fail to teach that the control zone contains a chemichromic dye.

Lawrence et al. teach a pH and amine test element that is useful in the diagnosis of vaginal infections. The test element contains a test section that is capable of detecting volatile amines in a test sample, wherein the amine test section contains an indicator that undergoes a detectable color change in the presence of a volatile amine.

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The amine test section also contains a second indicator that functions as a control and undergoes a detectable color change regardless of the presence of volatile amines in the test sample. The inclusion of a control that contains a color-changing indicator in both the pH and amine test sections of the test element is useful in order to assure that the indicator is not malfunctioning for reasons such as manufacturing error in the device, and that the device has been exposed to sufficient sample to produce a reading if the sample is indeed positive (see Figures; Abstract; column 4, lines 5-25 and lines 64-67; column 5, lines 1-17; column 9, lines 54-57; column 10, lines 20-27; column 11, lines 22-54; and column 12, lines 7-27).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to include with the device of Daniels et al., Miller et al. and Caillouette a control zone with a color-changing (chemichromic) dye as taught by Lawrence et al. because Lawrence et al. teach the benefit of including a control region with a second color-changing indicator, wherein a test area contains a first color-changing indictor, in order to assure that the indicator is not malfunctioning for reasons such as manufacturing error in the device, and that the device has been exposed to sufficient sample to produce a reading if the sample is indeed positive.

#### Conclusion

- No claims are allowed.
- 11. The following prior art made of record and not relied upon is considered pertinent to applicant's disclosure:

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Scaringe et al. (US 2003/0087451) teach an acid test kit utilizing various indicators, including alpha-naphtholbenzein, which produce readily detectable color

changes in the presence of various pH (see Abstract; and Table on pg 2).

Any inquiry concerning this communication or earlier communications from the examiner should be directed to JACQUELINE DIRAMIO whose telephone number is

(571)272-8785. The examiner can normally be reached on M-F 9-5:30.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mark Shibuya can be reached on 571-272-0806. The fax phone number for

the organization where this application or proceeding is assigned is 571-273-8300.

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/Jacqueline DiRamio/

Examiner, Art Unit 1641